Fire Forensics Investigation Day 3

October is Fire Prevention Month! Learn how fires start and spread through an interactive investigation this week!



INVESTIGATORS ACADEMY

_Investigators Academy



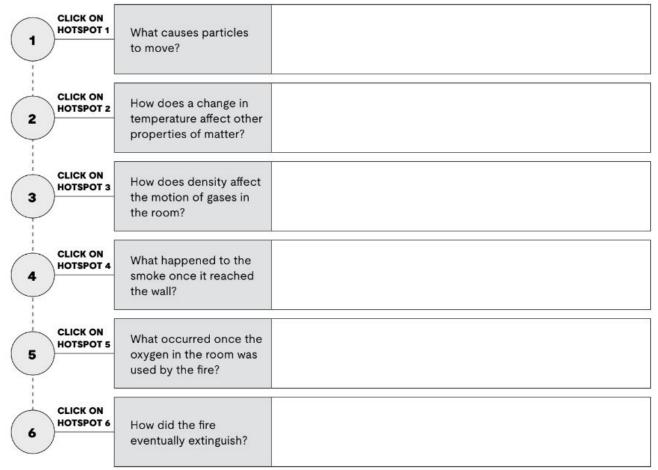
How Does Fire Behave Interactive

Scroll to the interactive where there is a **bed on fire.** Gather and communicate information from the interactive about how a fire behaves and interacts with matter.

How does fire behave?

1. Record your understanding of how fire behaves in the chart below.

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2. What is something new you learned about fire from the interactive?

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During a fire, changes in temperature affect the pressure and volume of gases in the room. These changes cause fire to behave in predicable ways.

Scroll to the red language accordion to better understand the stages of fire development.

- 3. Using information from the accordion, identify the best explanations for each term or core idea.
 - _____Temperature a. force exerted on walls of a container
 - b. amount of space consumed by an object
 - c. measurement of total kinetic energy of an object



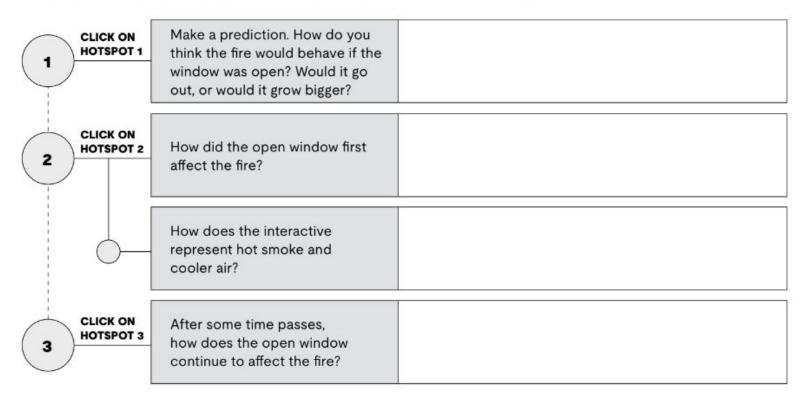
_ Pressure

Volume

How Does Ventilation Affect Fire Interactive

Scroll to the interactive where a **bed is on fire.** Gather and communicate information from the interactive about how ventilation affects fire interactions with matter.

4. Record your understanding of how ventilation affects fire in the chart below.



5. What is something new you learned about fire from the interactive?



Ventilation is the exchange of hot, oxygen-depleted gases, with cool, oxygen-rich air. Scroll to the red language accordion to better understand the stages of fire development.

6. Using information from the accordion, identify the best explanations for each term or core idea.

Exchange	a. fills lower pressure area
Hot, oxygen depleted gas	b. unequal pressure of inside hot air and outside cold air that creates a circulation of air
Cool, oxygen rich gas	c. flows outside the room



Real Lab Footage: Fuels

Scroll to the real lab footage of **organic versus synthetic fuels.** Gather information as you observe the video.

7. In the space below, record your observations about organic and synthetic fuels.

ORGANIC FUELS

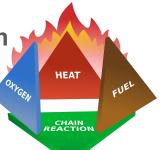
SYNTHETIC FUELS

Candle and Marshmallow Investigation:

How could we apply what we learned about particle movement, ventilation, and fuel types to our candle and marshmallow investigation. Draw a picture below.

Candle and Marshmallow Investigation

- 1. Observe your candle.
 - a. List the parts of the candle.



- 2. Light your candle. Observe closely. Use the screen to help block some light and visualize the structure of the flame and the wick of the candle.
 - a. Can you see space around the wick?
 - b. What is the fuel source for the flame since the solid wick isn't on fire? Blow your candle out and relight the fuel in the air to support this claim.
 - c. What happens when you put a piece of foil between the wick and the flame?
 - d. If you remove the fuel source, what happens to the fire?
- 3. Place the clear glass pyrex pan over the candle to test your predictions about how airflow or amount of oxygen affects fire behavior.
 - a. What happens to the fire if oxygen is depleted?
- 4. See if you can use ice in a beaker to "suck" the heat away from the reaction and put the fire out.
 - a. What happens to the fire if heat is removed?
- 5. Using the candle and the marshmallow on a toothpick explore the different types of heat transfer. Use a different marshmallow for each investigation.
 - a. Record your observations below

Conduction: heat transfer through touch or direct contact.	Convection: the transfer of heat from one place to another through the movement of a fluid or gas. (hot molecules rise and then fall through convection currents)	Radiation: the process of heat energy being transferred through space via electromagnetic waves.